

INDIAN SCHOOL AL WADI AL KABIR

First Rehearsal Examination 2022-23

SUB: Science (086)

Date: 11 /12/2022

Class: X

Time Allowed :3 hours

SET 1 Maximum Marks: 80

General Instructions:

- i. This question paper consists of 39 questions in 5 sections.
- ii. All questions are compulsory. However, an internal choice is provided in some questions. A student is expected to attempt only one of these questions.
- iii. Section A consists of 20 objective type questions carrying 1 mark each.
- iv. Section B consists of 6 Very Short questions carrying 02 marks each. Answers to these questions should in the range of 30 to 50 words.
- v. Section C consists of 7 Short Answer type questions carrying 03 marks each. Answers to these questions should in the range of 50 to 80 words
- vi. Section D consists of 3 Long Answer type questions carrying 05 marks each. Answer to these questions should be in the range of 80 to 120 words.
- vii. Section E consists of 3 source-based/case-based units of assessment of 04 marks each with sub-parts

	SECTION - A		
Select and write one most appropriate option out of the four options given for each of the questions $1 - 1$			
20			
Q. No	Questions	Marks	
1	Which of the following statements is correct about an aqueous solution of an acid	1	
	and a base?		
	I. Higher the pH, stronger the acid		
	II. Higher the pH, weaker the acid		
	III. Lower the pH, stronger the base		
	IV. Lower the pH, weaker the base		
	(a) I and II		
	(b) I and III		
	(c) I and IV		
	(d) II and IV		

2	Given below is a diagram showing an experimental set-up:	1
	Glass tube	
	home Inverted test-tube	
	Water containing	
	some acid	
	Cathode Anode	
	le sa	
	- Switch	
	6V	
	Identify the cases Y and V evolved during the experiment	
	(a) X is hydrogen and Y is oxygen	
	(b) X is oxygen and Y is hydrogen	
	(c) X is carbon dioxide and Y is oxygen	
	(d) X is chlorine and Y is hydrogen	
3	The heating of lead nitrate produces brown fumes of gas X. Identify gas X from	1
	the following.	
	(a) Lead oxide	
	(b) Oxygen	
	(c) Nitrogen dioxide	
	(d) Nitrogen	
4	A sample of soil is mixed with water and allowed to settle. The clear solution	1
	changes the pH paper yellowish orange. Which of the following would change the	
	(a) Lemon inice	
	(a) Lemon Juice (b) Vinegar	
	(c) Common salt	
	(d) An antacid	
5	On placing an iron nail in a test tube containing blue copper sulphate solution, it will	1
-	be observed that the copper sulphate solution	
	(a) Turns green, and a grey substance is deposited on the iron nail.	
	(b) Turns colourless and a grey substance is deposited on the iron nail.	
	(c) Turns colourless and a reddish-brown substance is deposited on the iron	
	nail.	
	(d) Turns green and a reddish-brown substance is deposited on the iron nail.	
6	Sodium hydrogen carbonate when added to hydrochloric acid, evolves a gas. Which	1
	of the following statements are true about the gas evolved?	
	I. It turns lime water milky.	
	II. It extinguisnes a burning splinter.	
	IV. It is reddish brown in colour	
	(a) L and II	
	(b) I and III	
	(c) II and III	
	(d) II and IV	

7	Covalent compounds	1
	(a) have high melting and boiling points	
	(b) are mostly soluble in water	
	(c) are formed between atoms of metals and non-metals	
	(d) are formed by the sharing of electrons.	
0	The phloen in plants are managed blo for	1
8	(a) transmentation of water	1
	(a) transportation of food	
	(b) transportation of vistor and food	
	(d) transportation of ovugan and earbon di ovide	
0	Identify the figure and its function	1
	Identity the right and its function -	
	(a) Neuron, filtration of waste	
	(b) Nephron, filtration of waste	
	(c) Neuron, transmission of impulse	
	(d) Nephron, transmission of impulse	
10	Observe the figure given below and identify the event and the organism	1
	$ \begin{array}{c} \\ \hline \\ $	
	(a) Hydra, Regeneration	
	(b) Hydra, fission	
	(c) Planaria, Regeneration	
11	(d) Fianafia, fission During your determination what is the probability that the offerning will be a sinfo	1
	(a) 50%	1
	(b) 100%	
	(c) It varies from person to person	
	(d) 47 to 49%	
12	Which of the two sets belong to the same trophic level?	1
	(a) Rabbit: Tiger	
	(b) Vulture: Rat	
	(c) Grasshopper: Hawk	
	(d) Frog: Lizard	

13	A student carries out an experiment and plots the V-I graph of three samples of	1	
	nichrome wire with resistances R_1 , R_2 and R_3 respectively. Which of the following is		
	true?		
	\mathbf{A} $\mathbf{R}_{\mathbf{I}}$		
	r / R_{2}		
	(ampere)		
	V(volts)		
	a. $\mathbf{K}_1 = \mathbf{K}_2 = \mathbf{K}_3$		
	$\mathbf{D} \cdot \mathbf{K}_1 > \mathbf{K}_2 > \mathbf{K}_3$		
	$\mathbf{C} \cdot \mathbf{K}_3 > \mathbf{K}_2 > \mathbf{K}_1$		
14	$\mathbf{d} \cdot \mathbf{K}_2 > \mathbf{K}_3 > \mathbf{K}_1$	1	
14	A student wants to obtain magnified image of an object AB as on a screen. Which	1	
	one of the following arrangements snows the correct position of AB for him/her to		
	De successiui?		
	(a) \xrightarrow{B} (b) \xrightarrow{B} (c)		
	\mathbf{c}_1		
	N N		
	$ \qquad \qquad$		
	(c) $(d) = B$		
	$2F$ F $2F_1$ F_1 C_1		
	V V V N N		
15	If the speed of light in vacuum is 3×10^8 m/s, find the speed of light in a medium	1	
	of absolute refractive index 1.5.		
	a) 2.14		
	b) 3×10^8 m/s		
	c) $2 \times 10^8 \text{ m/s}$		
	d) 4.5 x10 ⁸ m/s		
16	A cylindrical conductor of length 'l' and uniform area of cross-section 'A' has	1	
	resistance 'R'. The area of cross-section of another conductor of the same material and		
	same resistance but of length '21' is		
	a) A/2		
	b) 3A/2		
	c) 2A		
0.15	d) 3A		
Q. no 17 to 20 are Assertion - Reasoning based questions. These consist of two statements – Assertion			
(A) and Reason (R). Answer these questions selecting the appropriate option given below:			
(a) Both A and R are true and R is the correct explanation of A			
(b) Boin A and K are true and K is not the correct explanation of A			
(c) A is true but K is false			
(d) A is faise out K is true			
1/	Assertion(A): Calcium carbonate on nearing decomposes to give calcium oxide	1	
	and carbon dioxide $\mathbf{R}_{asson}(\mathbf{R})$: Photochemical decomposition occurs in the presence of surlight		
	Reason(R). I notoenemical accomposition occurs in the presence of sumght.		

18	Assertion (A): Amoeba shows multiple fission during unfavourable conditions.			1	
	Reason (R):	Reason (R): Chances of survival are less during unfavourable conditions.			
19	Assertion (A): The concentration of harmful chemicals is least in human beings.			1	
	Reason (R) : Man is at the apex of the food chain.				
20	Assertion (A): Hypermetropia is the defect of the eye in which only nearer objects			1	
	are seen.				
	Reason (R)	: Hypermetropia is corr	ected by using convex lens		
		SECTI	ON – B		
	I	Q. no. 21 to 26 are very	short answer questions		1
21	(a) Show the	e formation of MgO by th	e transfer of electrons.		2
	(b) Why are	e ionic compounds usually	y hard?		
	XX 71 . ·	OR I ALL I			
- 22	What is	cinnabar? How is metal e	extracted from cinnabar?	TT 1	2
22	In the follow	wing food chain, 100 J of	energy is available to the li	on. How much	2
	energy was	available to the producer	S?		
22	Plants $\rightarrow D$	$eer \rightarrow Lion$	ant modes of mutrition?		2
23	Define nutri	ition? What are the differ	ent modes of nutrition?		2
24	Draw a diag	gram of binary fission in A	Amoeda.	vince change of 1	2
25	Define 1 vo	rough a bettery of 3 V	of energy consumed in carr	ying a charge of 1	2
	coulonio un	OR			
	UK What is an ammater? A charge of 150 coulomb flows through a wire in one minute. Find				
	what is an ammeter? A charge of 150 coulomb flows through a wire in one minute. Find the electric current flowing through it				
26	How is the e	equal genetic contribution	of male and female parents	s ensured in the	2
	progeny?	- Jam Benerie control and			-
		SECTI	ON - C		
	(Q.no. 27 to 33 are short an	nswer questions.		
27	What is a re	dox reaction? When a ma	gnesium ribbon burns in air	with a dazzling	3
	flame and fo	orms a white ash, is magn	esium oxidised or reduced?	? Why?	
28	(a) Name the	e by-product of chlor-alka	li process which is used for	the manufacture	3
	of bleaching powder				
	(b) Explain why is hydrochloric acid a strong acid and acetic acid a weak acid.				
	(c) Name the compound which is obtained from baking soda and is used to				
•	remove peri	manent hardness of water			
29	In a cross between plants with round seeds and plants with wrinkled seeds the		3		
	offspring of F1 generation all had round seeds. When the F1 generation was self-				
	crossed, it was observed in the F2 generation that out of 100, 75 seeds were round.				
	Make a cross and answer the following:				
	(i) What is	the ratio of round: wrink	ed seeds in the F2 generation	nn?	
30	Analyse the	following observation tabl	e showing variation of image	-distance (v) with	3
50	object-distan	(u) in case of a convex	lens and answer the questions	s that follow	5
	without doing any calculations .				
	S.No	Object distance u (cm)	Image distance v (cm)		
	1	-100	+25		
	2	-60	+30		
	3	-40	+40		
	4	-30	+60		
	5	-25	+100		
	6	-15	+120		

(a) What is the focal length of the convex lens? Give reason to justify your answer. (b) Write the serial number of the observation which is not correct. On what basis have			
you arrived at this conclusion?			
(c) Select an appropriate scale and draw a ray diagram for the observation at S.No. 2.			
31 Name the type of mirror used in the following situations.	3		
(a) Headlights of a car			
(b) Side/rear-view mirror of a vehicle			
(c) Solar lufflace Support your answer with reason			
Support your answer with reason.			
32 Find the current flowing through the following electric circuit.	3		
3.6Ω 6Ω 3Ω			
10 30			
4.5 V			
OR			
Three resistors R_1 , R_2 and R_3 having values 5 Ω , 10 Ω , and 30 Ω respectively are			
connected in parallel across a battery of 12V. Calculate			
(a) the current in the circuit			
(c) the total circuit resistance.			
33 What will happen if all the carnivores are removed from the earth?	3		
OR			
What will happen to the grasslands if all the grazers are removed from there?			
O.no. 34 to 36 are Long answer questions.			
34 (a) Write the molecular formula of the following compounds and draw their	5		
electron dot structures.			
(i) Ethane (ii) Ethene			
(b) What are isomers? Write all the possible isomers of pentane.			
OR			
(a) Identify the functional group present in the following compounds			
н о			
(b) List any two characteristics of homologous series. Write the name and			
formula of the first member of alkene series.			
(c) Write the formula and electron dot structure of cyclopentane.			
35 (a) Draw a longitudinal section of a flower and label the following parts:	5		
 (1) Part that produces pollen grain. (ii) Part that transfers male generates to the female generates. 			
(iii) Part that is sticky to trap the pollen grain.			

	(iv) Part that develops into a fruit.	
	(b) Briefly explain double fertilization in plants.	
	OR	
	Draw a diagram of a diagram of a female reproductive system and label the part	
	(i) That produces egg	
	(ii) Where fusion of egg and sperm take place	
	(iii) Where zygote is implanted	
	(iv) The passage from where sperms enter	
	What happens to human egg when it is not fertilized?	
36	a. State Ohm's law	5
	b. With the help of a suitable circuit diagram prove that the reciprocal of the	
	equivalent resistance of a group of resistances joined in parallel is equal to the	
	sum of the reciprocals of the individual resistances.	
	c. In an electric circuit two resistors of 12 Ω each are joined in parallel to a 6 V	
	battery. Find the current drawn from the battery.	
	SECTION – E	
Q.no. 3	7 to 39 are case - based/data -based questions with 2 to 3 short sub - parts. Internal cho	oice is
provideo	l in one of these sub-parts.	
37	Ores mined from the earth are usually contaminated with large amounts of	4
	impurities such as soil, sand, etc., called gangue. The impurities must be removed	
	from the ore prior to the extraction of the metal. The processes Several steps are	
	involved in the extraction of pure metal from ores. Metals and Non-metals used	
	for removing the gangue from the ore are based on the differences between the	
	physical or chemical properties of the gangue and the ore. Different separation	
	techniques are accordingly employed.	
	i) Which of the following methods is used for the extraction of highly reactive	
	metals?	
	(a) Roasting	
	(b)Electrolytic reduction	
	(c) Electrolytic refining	
	(d)Calcination	
	ii) Why should the metal sulphides and carbonates be converted to metal oxides	
	in the process of extraction of metal from them?	
	iii) Differentiate between calcination and roasting with suitable examples.	
	OR	
	iii) How is copper obtained from its sulphide ore? Give equations of the reactions.	
38	As we have seen the process of sexual maturation is gradual, and takes place while	1
50	general body growth is still going on Therefore, some degree of sexual meturation	-7
	does not necessarily mean that the body or the mind is ready for sevual acts or for	
	having and bringing up children. How do we decide if the body or the mind is	
	ready for this major responsibility? All of us are under many different kinds of	
	reacy for this major responsionity: An of us are under many unferent kinds of pressures about these issues. There can be pressure from our friends for	
	pressures about mese issues. There can be pressure from our mends for	
	participating in many activities, whether we really want to or not. There can be	
	pressure from rammes to get married and start naving children. I here can be	
	pressure from government agencies to avoid naving children. In this situation,	
	making choices can become very difficult.	

	i. What is reproductive health?	
	ii. List any one kinds of pressures do we face in the society?	
	iii. Write any two importance of reproductive health?	
	OR	
	iii. Write any two preventive measures for STD?	
39	The twinkling of a star is due to atmospheric refraction of starlight. The starlight on entering the earth's atmosphere, undergoes refraction continuously before it reaches the earth. The atmospheric refraction occurs in a medium of gradually changing refractive index. Since the atmosphere bends starlight towards the normal, the apparent position of the star is slightly different from its actual position. The star appears slightly higher than its actual position when viewed near the horizon. Further this apparent position of the star is not stationary, but keeps on changing slightly, since the physical conditions of the earth's atmosphere are not stationary. Since the stars are very distant they are approximate point-sized source of light. As the path of rays of light coming from the star goes on varying slightly, the apparent position of the star fluctuates and the amount of starlight entering the eye flickers the stars sometimes appear brighter and at some other time, fainter, which is the twinkling effect. i) What is atmospheric refraction?	4
	ii) What will be the colour of sky as seen from the earth if there is no atmosphere?	
	iii) List two phenomena which can be explained on the basis of atmospheric refraction.	
	iii) Explain the reason why stars appear to twinkle and the planets do not twinkle.	